

UNIVERSITAS NEGERI YOGYAKARTA FACULTY OF MATHEMATICS AND NATURAL SCIENCES DEPARTMENT OF CHEMISTRY 1 Colombo Street Yogyakarta 55281 Phone (0274) 565411, Ext. 1398, Fax (0274)548203 Website: http://kimia.fmipa.uny.ac.id, E-mail: kimia@uny.ac.id

Bachelor of Science in Chemistry

MODULE HANDBOOK

Module name:	Chemical Process Industry							
Module level, if applicable:	Undergraduate							
Code:	KMA6317							
Sub-heading, if applicable:	•							
Classes, if applicable:	2							
Semester:	4 th							
Module coordinator:	Ir. Endang Dwi Siswani, M.T.							
Lecturer(s):	1. Ir. Endang Dwi Siswani, M.T.							
	2. Annisa Fillaeli, M.Si.							
Language:	English and bahasa Indonesia							
Classification within the	Compulsory Course							
curriculum:	400 minutes lestures 400 structured estivities and 400							
Teaching format / class	100 minutes lectures, 120 structured activities and 120							
hours per week during the	individual study per week							
semester: Workload:	Total workload is 01 hours not compater which consists of							
	Total workload is 91 hours per semester which consists of 100 minutes lectures, 120 structured activities and 120							
Cradit painta:	individual study per week for 16 weeks 2 SKS (3 ECTS)							
Credit points:								
Prerequisites course(s): Course Outcomes	After taking this course, the students have ability to:							
	CO1. Students are able to answer correctly about the scope of the Chemical Industry Process and How to manage the Chemical Industry safely, efficiently and effectively							
	CO2. Analyzing Process Engineering Flow Diagrams in a chemical industry							
	CO3. Students are able to explain correctly about the processes that occur in the Sugar Cane, Paper, Portland Cement industry							
	CO4. Students are able to explain correctly about the processes that occur in the Ammonia industry, Urea Fertilizers, Textiles, Milk Powder							
	CO5. Students are able to calculate correctly the amount of heat released during the direct sulfonation process							
	CO6. Students are able to correctly calculate the amount of							
	sulfur that has changed to							
	SO ₂ and SO ₃							
	CO7. Students are able to calculate correctly the amount of							
	oxygen needed so that the combustion of a kind of							
	hydrocarbon compound takes place perfectly							
Content: Chemical industrial process courses consist of the containing material on: Chapter I: Introduction, Chapter I: I								

	The production process in the Chemical Industry includes Industry: Sugar Cane, Paper, Petroleum, Portland Cement, Ammonia, Urea Fertilizer, Textile and Milk Powder, Chapter. III. Sulfonation Process, Chapter IV: Stoichiometry in Industry						
Study / exam achievements:	The final mark will be weight as follow:						
	No	СО	Assessment Object	Assessment Technique	Weight		
	1	CO1, CO2, CO3, CO4, CO5, CO6,	Individual assignment about observation reporting of waste treatment in the industry	Assignment	20%		
	2	CO7	PEFD analysis in certain industry	Assignment	20%		
	3		Mid term exam	Written test	30%		
	4		Final exam	Written test	30%		
				Total	100%		
Forms of media:	Board, LCD Projector, handouts, PPT slides, and stationaries						
Reference:	•						

PLO and CO mapping

	PLO1	PLO2	PLO3	PLO4	PLO5	PLO6	PLO7	PLO8	PLO9	PLO10
CO1	~									
CO2			✓							
CO3				✓						

CO4 CO5				\checkmark	
			√		
CO6					√
C07				\checkmark	