

UNIVERSITAS NEGERI YOGYAKARTA

FACULTY OF MATHEMATICS AND NATURAL SCIENCES DEPARTMENT OF CHEMISTRY

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Bachelor of Science in Chemistry

MODULE HANDBOOK

| Module name: | Selected Topics on Chemical Research | | | | | | |
|------------------------------------------------------------------------------------------------|------------------------------------------------------------|--|--|--|--|--|--|
| Module level, if applicable: | Undergraduate | | | | | | |
| Code: | KMA 6218 | | | | | | |
| Sub-heading, if applicable: | - | | | | | | |
| Classes, if applicable: | 2 | | | | | | |
| Semester: | 5 th | | | | | | |
| Module coordinator: | Jaslin Ikhsan, Ph.D | | | | | | |
| Lecturer(s): | 1. Jaslin Ikhsan, Ph.D | | | | | | |
| | 2. Dr. Sri Handayani | | | | | | |
| Language | 3. Dr. Cahyorini Kusumawardani | | | | | | |
| Language: Classification within the | English Compulsory Subject | | | | | | |
| | Compulsory Subject | | | | | | |
| curriculum: Teaching format / class | 100 minutes lectures, 120 structured activities and 120 | | | | | | |
| hours per week during the | individual study per week | | | | | | |
| semester: | individual study per week | | | | | | |
| Workload: | Total workload is 90,67 hours per semester which consists | | | | | | |
| Workload. | of 100 minutes lectures, 120 structured activities and 120 | | | | | | |
| | individual study per week for 16 weeks | | | | | | |
| Credit points: | 2 SKS (3 ECTS) | | | | | | |
| Prerequisites course(s): | - | | | | | | |
| Course Outcomes | After taking this course, the students are expected to be | | | | | | |
| able to: | | | | | | | |
| | CO1 Show responsibility in writing chemical | | | | | | |
| | research designs | | | | | | |
| | CO2 Apply scientific methods and principles of using | | | | | | |
| | Information and Communication Technology | | | | | | |
| | (ICT) for the purposes of storing, analyzing, | | | | | | |
| | processing, and collecting data in the fields of | | | | | | |
| | chemistry, research and industry | | | | | | |
| | CO3 Analyze the implications of the development or | | | | | | |
| | implementation of science and technology in | | | | | | |
| | accordance with their expertise based on | | | | | | |
| | rules, procedures and scientific ethics in order | | | | | | |
| | to produce ideas and solutions to chemical problems | | | | | | |
| | CO4 Writing chemical research ideas that show | | | | | | |
| | innovation for society | | | | | | |
| Content: | Selected Chemistry Research Topics study various | | | | | | |
| | research topics that have been developed and the ethics | | | | | | |
| | of writing articles: | | | | | | |
| | Definition of Ethics and Style of Article Writing in the | | | | | | |
| field of science, chemical aspects 2. Selection of themes from research articles in the field | | | | | | | |

| Forms of media: Har | of organic chemistry, inorganic chemistry, biochemistry, analytical chemistry, and physical chemistry from international and national journals. 3. Analysis of research articles on the fields of organic chemistry, inorganic chemistry, biochemistry, analytical chemistry and physical chemistry from international journals and national journals 4. Writing the Report of Analysis Results 5. Preparation of TAS Introduction Attitude assessment is carried out at each meeting by observation and/or self-assessment techniques using the assumption that basically every student has a good attitude. The student is marked very good or not good attitude if they show it significantly compared to other students in general. The result of attitude assessment is not taken into account in the final grades, but as one of the requirements to pass the course. Students will pass from this course if at least have a good attitude. The final mark will be weight as follow: | | | | | | | | |
|---------------------|------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------|---------------------------------------------------------------------------------------------------------|-----------------------------|-------------------------|--|--|--|--|--|
| Forms of media: Har | No CO Assessment Assessment Object Technique | | | | | | | | |
| | CO3, | a. Participationb. Assignmentc. Mid-term examd. Final Exam | Presentation / written test | 5% 25% 30% 40% | | | | | |
| + | Total 100% | | | | | | | | |
| 3. | Handout, Board, LCD Projector, Laptop/Computer, Module Ali Saukah dan Mulyadi Guntur Waseso (2006), Menulis Artikel untuk Jurnal Ilmiah, Universitas Negeri Malang, Malang. Mabry, T.J. (2001), Selected Topics from Forty Years of Natural Products Research: Betalains to Flavonoids, Antiviral Proteins, and Neurotoxic Nonprotein amino Acids, Journal of Natural Products, 64, 12, 1596-1604 Sing, I.S. (2015), The effect of using Concept Maps on Student Achievement in Selected Topic in Chemistry at Tertiary Level, Journal of Education and Practice. Nol 6, No 15 Tellez, F. (2020), 33 Critical Topics in Chemistry for | | | | | | | | |

PLO and CO mapping

| | PLO | | | | | | | | | |
|-----|----------|---------------|------|-----------|------|------|------|----------------|------|-------|
| CO | Attitude | General Skill | | Knowledge | | | | Specific Skill | | |
| | PLO1 | PLO2 | PLO3 | PLO4 | PLO5 | PLO6 | PLO7 | PLO8 | PLO9 | PLO10 |
| CO1 | | | | | | | | | | |
| CO2 | | | | | | | | | | |
| CO3 | | | | | | | | V | | |
| CO4 | | | | | | | | | | V |