



UNIVERSITAS NEGERI YOGYAKARTA
FACULTY OF MATHEMATICS AND NATURAL SCIENCES
DEPARTMENT OF CHEMISTRY
1 Colombo Street Yogyakarta 55281
Phone (0274) 565411, Ext. 1398, Fax (0274)548203
Website: <http://kimia.fmipa.uny.ac.id>, E-mail: kimia@uny.ac.id

Bachelor of Science in Chemistry

MODULE HANDBOOK

Module name:	Structure Analysis of Inorganic Compounds
Module level, if applicable:	Undergraduate
Code:	KMA 6224
Sub-heading, if applicable:	-
Classes, if applicable:	-
Semester:	6 th
Module coordinator:	Prof. A. K. Prodjosantoso, Ph.D.
Lecturer(s):	Prof. A. K. Prodjosantoso, Ph.D.
Language:	Bahasa Indonesia, English
Classification within the curriculum:	Elective Course
Teaching format / class hours per week during the semester:	Lectures: 100 minutes lectures, 120 structured activities and 120 individual study per week
Workload:	Total workload of the activity is 90,67 hours per semester which consists of 100 minutes lectures, 120 structured activities and 120 individual study per week for 16 weeks
Credit points:	2 SKS (3 ECTS)
Prerequisites course(s):	-
Course Outcomes	After taking this course, the students have ability to: CO1. Able to define the d^n for transition metal in coordination compound and predict the type of its transition CO2. Able to apply the Tanabe-Sugano curve to analyze the UV-Vis spectra of coordination compound and

	<p>define the relation between the UV-Vis spectra character towards structure</p> <p>CO3. Able to define the band gap energy and optic properties of inorganic compound from its UV-Vis Spectra</p> <p>CO4. Able to interpret the FTIR spectra of inorganic compound</p> <p>CO5. Able to make relation, define, compare, and evaluate FTIR of aluminosilicate materials toward its chemical structure</p> <p>CO6. Able to analyze XRD diffractogram and describe the relation between the parameters and the structure of crystalline materials</p> <p>CO7. Able to define the relation between X Ray diffraction pattern with FTIR spectra data</p> <p>CO8. Able to analyze the microstructure of inorganic materials from SEM graph</p> <p>CO9. Able to do a search and describe the results of their study using their own language regarding the research in structure analysis of inorganic material</p>																									
Content:	This course discusses elucidation of inorganic materials using UV-Vis Spectrometer, FT-IR, X-Ray Diffraction (XRD), and SEM instruments. The study focuses on the relation between the structure and properties of inorganic salts, coordination compound, aluminosilicate materials (zeolite, clay), and metal oxide.																									
Study / exam achievements:	<p>The final mark will be weight as follow:</p> <table border="1"> <thead> <tr> <th>No</th> <th>CO</th> <th>Assessment Object</th> <th>Assessment Technique</th> <th>Weight</th> </tr> </thead> <tbody> <tr> <td>1</td> <td>CO1, CO2, CO3, CO4, CO5,</td> <td>Structural assignment: ability to rasonalize and describing</td> <td>Assignment</td> <td>15%</td> </tr> <tr> <td>2</td> <td>CO6, CO7, CO8, CO9</td> <td>Structural assignment: ability to applying the formula according to context</td> <td>Assignment</td> <td>15%</td> </tr> <tr> <td>3</td> <td></td> <td>Structural assignment: ability to collaborate, analyze, rasonalize, and communicate</td> <td>Assignment</td> <td>15%</td> </tr> <tr> <td>4</td> <td></td> <td>Individual assignment: skill to collect literacy,</td> <td>Assignment</td> <td>15%</td> </tr> </tbody> </table>	No	CO	Assessment Object	Assessment Technique	Weight	1	CO1, CO2, CO3, CO4, CO5,	Structural assignment: ability to rasonalize and describing	Assignment	15%	2	CO6, CO7, CO8, CO9	Structural assignment: ability to applying the formula according to context	Assignment	15%	3		Structural assignment: ability to collaborate, analyze, rasonalize, and communicate	Assignment	15%	4		Individual assignment: skill to collect literacy,	Assignment	15%
No	CO	Assessment Object	Assessment Technique	Weight																						
1	CO1, CO2, CO3, CO4, CO5,	Structural assignment: ability to rasonalize and describing	Assignment	15%																						
2	CO6, CO7, CO8, CO9	Structural assignment: ability to applying the formula according to context	Assignment	15%																						
3		Structural assignment: ability to collaborate, analyze, rasonalize, and communicate	Assignment	15%																						
4		Individual assignment: skill to collect literacy,	Assignment	15%																						

			understanding, and describing		
	5		Mid term exam	Written test	20%
	6		Final exam	Written test	20%
	Total				100%
Forms of media:	Board, LCD Projector, handouts, PPT slides, and stationaries				
Reference:	<p>A. Ángel Vegas Molina, 2018, Structural Models of Inorganic Crystals. From the Elements to the Compounds, 1st ed., Universitat Politècnica de València</p> <p>B. Jan Chyba, Martin Novák, Petra Munzarová, Jan Novotný, and Radek Marek, 2018, Through-Space Paramagnetic NMR Effects in Host–Guest Complexes: Potential Ruthenium(III) Metallodrugs with Macrocyclic Carriers, <i>Inorganic Chemistry</i>, 57, 15, 8735-8747</p> <p>C. Lukáš Jeremias, Jan Novotný, Michal Repisky, Stanislav Komorovsky, and Radek Marek, 2018, Interplay of Through-Bond Hyperfine and Substituent Effects on the NMR Chemical Shifts in Ru(III) Complexes, <i>Inorganic Chemistry</i>, 57, 15, 8748-8759</p> <p>D. Longfei Li, Ming Lei, Li Liu, Yaoming Xie, and Henry F. Schaefer III, 2018, Metal–Substrate Cooperation Mechanism for Dehydrogenative Amidation Catalyzed by a PNN-Ru Catalyst, <i>Inorganic Chemistry</i>, 57, 15, 8778-8787</p> <p>E. Muller U (2008), <i>Inorganic Structural Chemistry</i>, 2nd Edition, Wiley VCH</p> <p>F. Sangeeta, D., LaGraff, J.R. (2004), <i>Inorganic Materials Chemistry Desk Reference</i>, 2nd Edition, CRC Press</p> <p>G. Setia Budi, A (2007), <i>Karakterisasi Material: Prinsip dan Aplikasinya dalam Penelitian Kimia</i>, Erlangga</p> <p>H. Suminar, S (1999), <i>Analisis Data Difraksi dengan Metode Rietveld</i>, Pustaka Mandiri</p>				

PLO and CO mapping

CO	PLO									
	Attitude	Generic Skills		Knowledge				Specific Skills		
	PLO1	PLO2	PLO3	PLO4	PLO5	PLO6	PLO7	PLO8	PLO9	PLO10
CO1					✓					
CO2					✓					
CO3					✓					
CO4							✓			
CO5							✓			
CO6					✓					
CO7					✓					
CO8									✓	
CO9					✓					