



UNIVERSITAS NEGERI YOGYAKARTA
 FACULTY OF MATHEMATICS AND NATURAL SCIENCES
 DEPARTMENT OF CHEMISTRY
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Bachelor of Science in Chemistry

MODULE HANDBOOK

Module name:	Material Chemistry
Module level, if applicable:	Undergraduate
Code:	KMA6228
Sub-heading, if applicable:	-
Classes, if applicable:	-
Semester:	6 th
Module coordinator:	Prof. AK Prodjosantoso
Lecturer(s):	Dr Cahyorini Kusumawardani
Language:	English
Classification within the curriculum:	Elective Course
Teaching format / class hours per week during the semester:	Lectures: 100 minutes lectures, 120 structured activities and 120 individual study per week
Workload:	Total workload of the activity is 90,67 hours per semester which consists of 100 minutes lectures, 120 structured activities and 120 individual study per week for 16 weeks
Credit points:	2 SKS (3 ECTS)
Prerequisites course(s):	-
Course Outcomes	<p>After taking this course, the students have ability to:</p> <p>CO1. Able to describe the concept and terminology of material chemistry, history and development of material chemistry, material classification, and material application</p> <p>CO2. Able to explain the material structure, crystallography, bonding, synthesis and processing, thermodynamics, and kinetics</p> <p>CO3. Able to describe metal, metal oxide, alloys, conductor, semiconductor, and isolator</p> <p>CO4. Able to explain ceramics and glasses material</p> <p>CO5. Able to describe polymer materials, advanced polymer materials: synthesis and application</p> <p>CO6. Able to explain composites material and biomaterial</p> <p>CO7. Able to explain liquid crystal materials: thermo trophic, calamite, metal organic, dichotic, and polymer</p> <p>CO8. Able to explain nanostructure and nanoparticle materials, and the related analysis</p> <p>CO9. Able to explain porous and layer materials, and the related analysis</p> <p>CO10. Able to do a search and describe the results of their study using their own language regarding the research in material and its application</p>
Content:	This course discusses various aspect of chemistry in the design and discovery of new as an important role to

	synthesis future functional materials. It advance understanding how the history of a material influences its structure, properties, and performance. Lecture emphasizes the mastery of lecture material logically and scientifically and the ability to use scientific methods to solve problems faced by students																																					
Study / exam achievements:	<p>The final mark will be weight as follow:</p> <table border="1"> <thead> <tr> <th>No</th> <th>CO</th> <th>Assessment Object</th> <th>Assessment Technique</th> <th>Weight</th> </tr> </thead> <tbody> <tr> <td rowspan="5">1</td> <td rowspan="5">CO1, CO2, CO3, CO4, CO5, CO6, CO7, CO8, CO9, CO10</td> <td>Attitude</td> <td>Observation</td> <td>20%</td> </tr> <tr> <td>Structural assignment: ability to rasonalize and describing</td> <td>Assignment</td> <td>10%</td> </tr> <tr> <td>Structural assignment: ability to applying the formula according to context</td> <td>Assignment</td> <td>10%</td> </tr> <tr> <td>Structural assignment: ability to collaborate, analyze, rasonalize, and communicate</td> <td>Assignment</td> <td>10%</td> </tr> <tr> <td>Individual assignment: skill to collect literacy, understanding, and describing</td> <td>Assignment</td> <td>10%</td> </tr> <tr> <td></td> <td></td> <td>Mid term exam</td> <td>Written test</td> <td>20%</td> </tr> <tr> <td></td> <td></td> <td>Final exam</td> <td>Written test</td> <td>20%</td> </tr> <tr> <td></td> <td></td> <td>Total</td> <td></td> <td>100%</td> </tr> </tbody> </table>	No	CO	Assessment Object	Assessment Technique	Weight	1	CO1, CO2, CO3, CO4, CO5, CO6, CO7, CO8, CO9, CO10	Attitude	Observation	20%	Structural assignment: ability to rasonalize and describing	Assignment	10%	Structural assignment: ability to applying the formula according to context	Assignment	10%	Structural assignment: ability to collaborate, analyze, rasonalize, and communicate	Assignment	10%	Individual assignment: skill to collect literacy, understanding, and describing	Assignment	10%			Mid term exam	Written test	20%			Final exam	Written test	20%			Total		100%
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		Final exam	Written test	20%																																		
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Forms of media:	Board, LCD Projector, handouts, PPT slides, and stationaries																																					
Reference:	<p>Hojo, J. (2019). <i>Materials chemistry of ceramics</i>. Singapore: Springer Nature Singapore</p> <p>Fahlman, B. D. (2011). <i>Maerials chemistry 2nd ed</i>. Mount Pleasant, MI. USA: Springer</p> <p>Lawrence, HV (1992), <i>Elements of Material Science and Engineering</i>, Pearson Education</p> <p>Allock, H. R. (2019). <i>Inroduction to materials chemistry 2nd edition</i>. Wiley</p> <p>Lawrence, HV (1996), <i>Elements of Material Science</i>, Addison-Wesley Longman,</p> <p>Burns, G.; Glazer, A.M. (1990). <i>Space Groups for Scientists and Engineers (2nd ed.)</i>. Boston: Academic Press, Inc</p> <p>Lee, J.Y., Farha, O.K., Roberts, J., Scheit, K.A., Nguyen, S.T., and Hupp, J.T. (2019), Metal-organic framework materials as catalyst, <i>Chemical Society Reviews</i>, 5, 144-192</p>																																					

PLO and CO mapping

CO	PLO									
	Attitude	Generic Skills		Knowledge				Specific Skills		
	PLO1	PLO2	PLO3	PLO4	PLO5	PLO6	PLO7	PLO8	PLO9	PLO10
CO1							✓			
CO2					✓					
CO3					✓					
CO4					✓					
CO5					✓					
CO6					✓					
CO7					✓					
CO8									✓	
CO9									✓	
CO10									✓	

