

## **UNIVERSITAS NEGERI YOGYAKARTA**

FACULTY OF MATHEMATICS AND NATURAL SCIENCES DEPARTMENT OF CHEMISTRY

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## **Bachelor of Science in Chemistry**

## **MODULE HANDBOOK**

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Module level, if applicable:	Undergraduate						
Code:	KMA 6234						
Sub-heading, if applicable:	-						
Classes, if applicable:	-						
Semester:	6 <sup>th</sup>						
Module coordinator:	Prof. Dr. Nurfina Aznam, Apt.S.U						
Lecturer(s):	Prof. Dr. Nurfina Aznam, Apt.S.U						
Language:	Bahasa Indonesia						
Classification within the	Elective Course						
curriculum:							
Teaching format / class	Lectures: 100 minutes lectures, 120 structured activities and						
hours per week during the	120 individual study per week						
semester:	, .						
Workload:	Total workload of the activity is 90,67 hours per semester						
	which consists of 100 minutes lectures, 120 structured						
	activities and 120 individual study per week for 16 weeks						
Credit points:	2 SKS (3 ECTS)						
Prerequisites course(s):	-						
Course Outcomes	After taking this course, the students have ability to:						
	CO1. explain about isolation and identification of active						
	compounds in plants that have been used empirically						
	for treatment						
	CO2. Explain about synthesis of analogous structures from						
	basic forms of compounds which have potential						
	treatment activities.						
	CO3. describe how to finding new parent structures by means						
	of synthesis of organic compounds, with or without						
	contact with natural active substances						
	CO4. Explain about the Linking of the chemical structure with						
	the way drugs work. CO5. Explain about how develop drug designs						
	CO6. Explain about how develop the relationship of chemical						
	structures and biological activities through physical						
	chemical properties with the help of statistics.						
Content:	This course studies about the working relationship of drugs,						
	the relationship between chemical structure and biological activity of biodynamics through physical properties and						
	chemical reactivity of compounds.						
Study / exam achievements:	The final mark will be weight as follow:						
	No CO Assessment Assessment Weight						
	Object Technique						

	1	CO1,	Attitude and	Observation			
	I			Observation			
		CO2,	activity	Aggiggages	1		
	CO3, CO4,		Structural	Assignment,			
			assignment:	Presentation,			
		CO5,	ability to	discussion			
		CO6.	rasionalize and				
			describing Structural	2004			
			ability to		30%		
			applying the formula				
		according to context					
			<u> </u>				
			Structural				
			assignment:				
			ability to				
			collaborate,				
			analyze, rasionalize, and				
			communicate				
			Individual	Assignment	-		
			assignment:	Assignment			
			skill to collect				
			literacy,				
			understanding,				
			and describing				
			Mid term exam	Written test	30%		
			Final exam	Written test	40%		
		<u> </u>	T IIIai Cxaiii	Total	100%		
Forms of media:	Boar	d I CD	Projector, Video, ha				
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Reference:	A. Ro	osensto	ck J., Bajaj H.S., Ja	anež A., et al. 20	20. Once-		
			nsulin for Type 2				
			eatment.N Engl J Me				
			an X, Wu P, et al				
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	dynamics in Wuhan, China, of novel coronavirus-infected pneumonia. N Engl J Med, 382, 1199-1207.						
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		_	Hu S, Gao J. 2020,		•		
			us disease 2019 (CC	OVID-19), Drug Dis	scov Ther,		
	<ul> <li>14, 58-60.</li> <li>D. Siswandono, S., 2016, Kimia Medisinal, Edisi-2, Airlangga University Press</li> <li>E. Ekinci, D, et al., 2012, Medicinal Chemistry and Drug Desain, Published by InTech.</li> <li>F. Thomas, G., 2003, Fundamentals of Medicinal Chemistry, John willey and Sons Ltd</li> </ul>						

## **PLO and CO mapping**

	PLO									
СО	Attitude	Generic Skills		Knowledge				Specific Skills		
	PLO1	PLO2	PLO3	PLO4	PLO5	PLO6	PLO7	PLO8	PLO9	PLO10
CO1						✓		✓		
CO2						✓		✓		
CO3						✓		✓		
CO4						✓		✓	✓	
CO5						✓		✓	✓	
CO6						✓		✓	✓	