



UNIVERSITAS NEGERI YOGYAKARTA
 FACULTY OF MATHEMATICS AND NATURAL SCIENCES
 DEPARTMENT OF CHEMISTRY
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Bachelor of Science in Chemistry

MODULE HANDBOOK

Module name:	Statistics
Module level, if applicable:	Undergraduate
Code:	MKU6210
Sub-heading, if applicable:	-
Classes, if applicable:	2
Semester:	1 st
Module coordinator:	Endang Listyani, M.Si
Lecturer(s):	Endang Listyani, M.Si
Language:	Bahasa Indonesia
Classification within the curriculum:	Compulsory Course
Teaching format / class hours per week during the semester:	100 minutes lectures, 120 structured activities and 120 individual study per week
Workload:	Total workload is 90,67 hours per semester which consists of 100 minutes lectures, 120 structured activities and 120 individual study per week for 16 weeks
Credit points:	2 SKS (3 ECTS)
Prerequisites course(s):	-
Course Outcomes	<p>After taking this course, the students have ability to:</p> <p>CO1. Show piety, politeness and ethics during the learning process</p> <p>CO2. Use IT to analyse research data in chemistry</p> <p>CO3. Employ hypotesis testing to analyse research data in chemistry</p> <p>CO4. Be independent, responsible and cooperative during the learing process</p> <p>CO5. Explain the function of statistics to solve problems in a daily life</p> <p>CO6. Define and systematically evaluate the data description and frequency distribution to bring innovation in analyzing the reserach data on chemistry</p> <p>CO7. Implement aritmatic mean and geometrics, median and mode to solve problems related to chemistry</p> <p>CO8. Calculate data variation (rage, standardeviation and variant) and probability to solve problems related with chemistry</p> <p>CO9. Explain the concept of hypothesis testing and read kinds of distribution tables to solve problems related to chemistry</p>
Content:	This course discusses the basic concept of statistics, data description, probability, probability distribution, hypotesis testing, and samples of intepretation.

Study / exam achievements:	The final mark will be weight as follow:				
	No	CO	Assessment Object	Assessment Technique	Weight
	1	CO3, CO4, CO5 and CO9	Individual study	Tasks	10%
	2	CO4 and CO9	Structured activities	Observing presentation skills	20%
	3	CO1 and CO2	Observed attitude	Observing attitude	10%
	4	CO5, CO6, CO7, and CO8	Mid term	Written test	30%
	5	CO3 to CO9	Final term	Written test	30%
Total				100%	
Forms of media:	Board, LCD Projector, handouts, PPT slides, and stationaries				
Reference:	<p>Stevens, J. P. (2002). <i>Applied multivariate statistics for the social sciences</i>. New Jersey: Routledge</p> <p>Rumsey, D. J. (2016). <i>Statistics for dummies 2nd edition</i>. United States: The Ohio State University</p> <p>Reinhart, A. (2015). <i>Statistics done wrong: The woefully complete guide</i>. No Strach Press</p> <p>Frost, J. (2020). <i>Introduction to statistics: An intuitive guide for analyzing data and unlocking discoveries</i>. Jim Publishing</p> <p>Walpole, R. E. (1995). <i>Pengantar statistika edisi ke-3</i>. Translated by Bambang Sumantri. Jakarta: Gramedia</p> <p>Kirk, R. E. (1968). <i>Experimental design: Procedures for the behavioral sciences</i>. United States of America: International Thomson Publishing Inc.</p> <p>Underwood, B. J., Duncan, C. P., Taylor, J. A., & Cotton, J. W. (1954). <i>Elementary statistics</i>. East Norwalk, CT, US: Appleton-Century-Crofts.</p>				

PLO and CO mapping

	PLO1	PLO2	PLO3	PLO4	PLO5	PLO6	PLO7	PLO8	PLO9	PLO10
CO1					✓					
CO2					✓					
CO3					✓					
CO4							✓			
CO5							✓			
CO6							✓			
CO7										✓
CO8										✓
CO9										✓